TASK 5

## INSTRUCTIONS

Hello 6th graders! Follow these steps:

- Step I: Watch and listen to this video:
https://drive.google.com/file/d/InIZ6hZOHRX_r7DTmJnr7VDKPsdI98Egp/view?usp=sharing
- Step 2 : Read the PWP slides again.
- Step 3: Do UNIT 4 lesson 3 and 4 activities' of your online Activity Book - Module 2.

$$
\begin{aligned}
& \text { WHAT IS A } \\
& \text { MIXTURE? }
\end{aligned}
$$

## A MIXTURE IS...

A substance made up of two or more types of matter that aren't chemically combined.

The components of a mixture can be easily SEPARATED.
$\checkmark$ The components each keep their ORIGINAL PROPERTIES.
The quantity of the components is variable.

There are two main categories of mixtures.

## MXXTURES



## MXXTURES



## CONSISTENT colour and

texture.

VARIED
colour and
texture.


## Are they HOMOGENEOUS



## or HETEROGENEOUS mixtures?



## HOMOGENEOUS



## HETEROGENEOUS

$$
\begin{aligned}
& \text { WHAT IS A } \\
& \text { SOLUTION? }
\end{aligned}
$$

## A SOLUTION IS...

A solution is a mixture of one or more liquids with one or more substances that are dissolved in the liquid.

SOLVENT: the substance that dissolves.
SOLUTE: the dissolved substance.

## SOLUTE



SOLVENT

$$
\begin{aligned}
& \text { SEPARATING } \\
& \text { MIXTURES }
\end{aligned}
$$

## FILTRATION

is used to separate an insoluble solid from a mixture.


As the mixture passes through the filter, the solid particles are trapped in it.


EVAPORATION
is used to separate a soluble solid from a liquid.


## EVAPORATION



## EVAPORATION



DISTILLATION
is used to separate a liquid from a mixture of two or more liquids.


DISTILLATION


DISTILLATION

## DISSOLUTION

is used to separate a soluble solid from an insoluble one.

Eg. Sand and salt mixture

To separate a mixture using dissolution, follow these steps.

I. Add water to the mixture to dissolve the soluble solid.
2. Filter the mixture to separate the soluble solid from the insoluble one.
3. Evaporate the remaining solution to separate the soluble solid from the water.

## © THANKS!

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